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1 Flammability trends for a comprehensive array of cladding materials

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10

11 Highlights:

- 12 • Organic content is shown to correlate poorly with fire performance
- 13 • Residue in an inert atmosphere may be useful for rapid screening purposes
- 14 • Cladding materials within generic categories behave very differently
- 15 • Specific cladding materials with potential low flame spread are highlighted

16

17 Abstract:

18 The flammability of materials is a key component of modern cladding fires. Vertical flame
19 spread is a complex phenomenon which is, amongst others, a function of thermal inertia, ignition
20 temperature, and heat release. The recently published Cladding Materials Library contains the
21 needed flammability data to help engineers perform fire risk assessments on buildings. Cladding
22 fire research has previously generally focused on expensive and time consuming full-scale
23 testing, or on the chemical composition, with little regard to the flammability or other
24 performance metrics. Here we show common trends in the ignition and burning behaviour for
25 cladding materials in a systematic bench-scale study using the Cladding Material Library. The
26 organic content is shown to be a poor indicator of the fire performance, as represented by the
27 heat release rate. A simple and highly conservative model indicates the relative behaviour of a
28 diverse range of cladding materials. This analysis supports competent engineers to select which
29 specific buildings require further investigation based on performance, and to aid development of
30 remediation solutions. The differences within categories of materials, e.g. high-pressure
31 laminates, are large and thus the performance should be tailored for the specific building
32 material. This work complements but does not replace full-scale system testing.

33

34 **Keywords:** modeling; risk assessment; performance-based design; flame spread; fire growth;
35 ignition; heat release rate; hazard evaluation; façade fires

36

37 1. Introduction

38 Cladding fires are a modern problem which challenge the classic fire safety strategy. The
39 traditional fire safety strategy in buildings relies on the concept that there is no vertical spread of
40 fire. This was typically prevented through the use of vertical compartmentation and non-
41 combustible materials to prevent upward or downward spread of flames. Fire spread between
42 neighbouring buildings is also heavily influenced by the material selection in a façade system.
43 Within a building, different levels are also connected by stairwells which are heavily protected to
44 ensure safe egress and prevent the spread of smoke.

45 The use of modern materials and systems has challenged that strategy. The introduction of
46 flammable materials whose behaviour is not quantified or integrated into the strategy therefore
47 represents an unknown and unaccounted for fire risk. The use of these materials globally has
48 been rampant across the last 2–3 decades with little to no consideration for how the fire strategy
49 of a building is influenced by this. The Queensland Government requires the investigation of all
50 buildings which were built or renovated after 1994 [1].

51 Aluminium composite panels (ACPs) have been receiving much focus in the cladding fire
52 challenge, and the remediation and mitigation efforts have been heavily concentrated on these
53 materials. Fundamentally, ACPs are the symptom of a larger problem, and the fire risk of other
54 flammable components within a cladding must equally be assessed. These include high-pressure
55 laminates, timber claddings, weatherproof membranes, and insulation. The problem is
56 particularly difficult as these materials are included as part of a system which has interactions
57 between components, and introduces the issue of elements such as mechanical fixings. Many of
58 the remediation strategies around the world have yet to address whether or not non-ACP
59 materials pose a risk for the specific buildings they are included within.

60 In Queensland, a framework [2] was developed to deliver the needed flammability data to fire
61 engineers who were tasked with performing initial fire risk assessments for buildings. This
62 framework arms engineers with the basic data needed to make assessments for the remediation
63 and investigation of existing buildings. The data collected from all publicly-owned buildings in
64 Queensland, Australia is published in the Cladding Materials Library [3], and contains the
65 chemical composition, thermal degradation, heat of combustion, ignition characteristics, burning
66 behaviour, and flame spread. The complexity of the database is such that guidance is provided to
67 practitioners to describe its usage and interpretation [4] so that the data is used correctly.

68 This has resulted in an extensive database of common cladding materials, and gives insight into
69 the performance of the materials being used in actual buildings. A plethora of data is available to
70 be analysed to assess cladding materials and their relative performance. This acts as a first step to
71 aid choices in remediation and investigation, but it does not substitute full-scale testing. The goal
72 is to enable rational decision-making, as all combinations of systems cannot realistically be
73 tested at full-scale and sophisticated models have not been validated in this space.

74 The aim is to understand the ignition and flammability of common cladding materials, and
75 whether there are performance metrics which can be used. To achieve this, materials encountered
76 in the Cladding Materials Library [3] are studied and the key trends between different materials
77 are noted. This is used to assess whether a commonly used chemical composition metric in
78 Australia is appropriate for assessing cladding materials. A simple model is applied
79 conservatively to help gain insight into the potential relative performance of different materials.

80 2. Materials

81 The materials in this project are taken from publicly owned buildings in Queensland, Australia.
82 A total of 1,091 samples were taken, of which detailed flammability testing was performed on 20
83 materials. These materials are all published in the Cladding Materials Library [3]. The sample
84 preparation techniques and methods used to test the samples are described briefly below, and a
85 full description [5] and an examination of the sensitivities [6] can be found elsewhere.

86 Samples were initially either taken in the form of 40 mm diameter discs removed from buildings
87 using a hole-saw, or as $1 \times 1 \text{ m}^2$ sheets taken from buildings. The encapsulation for all samples
88 was removed, and testing was focused solely on the core material. This was so that the
89 fundamental material behaviour could be assessed, without the added complications of
90 composite effects which would be present in full products. This means that the aluminium skin
91 was removed from ACPs, and the metal sheet was removed from insulation sandwich panels.
92 The fire behaviour of products is an added level of complexity which can be studied later, and is
93 outwith the scope of this work.

94

95 3. Methods

96 Thermogravimetric Analysis (TGA) was performed using a Netzsch STA449 F3 Jupiter (ISO
97 11358-1 [7]). A heating rate of $20 \text{ }^\circ\text{C min}^{-1}$ was used in a range from $50\text{--}800 \text{ }^\circ\text{C}$, with one test in
98 air and one test in nitrogen, each with a gas flowrate of 150 ml min^{-1} . Sample mass was 10.1 ± 0.4
99 mg (error as standard deviation across 1,238 tests) for ACPs, and the mass was reduced for light
100 weight insulations or materials where insufficient mass could be collected, e.g. adhesives.
101 Alumina (Al_2O_3) crucibles with a volume of $70 \text{ }\mu\text{L}$ and no lid were used. Results are in the form
102 of mass and mass loss as a function of temperature, and the mass residue.

103 Quantitative material identification for all the materials presented in this paper has been
104 performed [5,6] but is not presented here.

105 The heat of combustion was obtained using bomb calorimetry (ISO 1716 [8]). Three trials were
106 performed in a Parr Instruments Calorimetric Thermometer Model 6772, and the result presented
107 as an average and standard deviation. For materials that did not readily ignite, a material with a
108 known heat of combustion – benzoic acid – was used to ensure ignition and that sufficient energy
109 was released. The heat contribution of the benzoic acid was then removed afterwards.

110 The critical heat flux was determined using a conical heater supplied by Fire Testing Technology
111 (East Grinstead, UK). This was according to Annex H of ISO 5660 [9] but without gas analysis
112 or a load cell. Samples of $100 \times 100 \text{ mm}^2$ were exposed to a range of heat fluxes until there was
113 at least one case with ignition and one case of no ignition, separated by at most 1.0 kW m^{-2} .

114 Flammability testing was performed in an ISO 5660 cone calorimeter [9] supplied by Fire
115 Testing Technology (East Grinstead, UK). Heat fluxes of 35, 50 and 60 kW m^{-2} were used with
116 two repetitions of each. In some cases 80 kW m^{-2} was used to give a wider spread of results. A
117 wire mesh with a metal retaining frame was used in all tests due to the expansion of many
118 materials, and the standard was otherwise followed. Ceramic wool was used as insulation at the
119 rear surface, and replaced between tests. The results presented are the heat release rate as a
120 function of time, mass as a function of time, total energy released, time to ignition, peak heat

121 release rate, and the mass residue. Additionally, the ignition temperature, total heat transfer
 122 coefficient at ignition, and apparent thermal inertia were calculated using the method outlined by
 123 Long *et al.* [10].

124

125 4. General heat release results

126 The Cladding Materials Library is currently in the form of a database which is free and open to
 127 access, which means any materials referred in this manuscript can be referred to at any time. The
 128 materials must be summarised in some form to aid discussion, analysis and comparisons. The
 129 materials have been grouped into basic categories where there is a degree of commonality (Table
 130 1) – for example, charring materials with a specific physical form – but the behaviour within
 131 these categories is still wide. The intention of the summary then is to aid discussion and is not a
 132 classification. If the materials are to ultimately be classified, then this should be done based on
 133 identified performance metrics which are deemed suitable. The samples IDs refer to aluminium
 134 composite panels (ACP_{xx}), insulation (INS_{xx}), sarking or weatherproof membranes (SRK_{xx}), or
 135 “other” materials (OTH_{xx}).

136 The descriptions in this manuscript generally do not refer to the full chemical composition as it is
 137 not needed for the level of comparisons made here. Nonetheless, the chemical compositions are
 138 all available in the database [3].

139 The heat release rate as a function of time gives an indication of the relative and distinct
 140 behaviours of different cladding materials (Fig. 1). A short description of each category is given,
 141 indicating the general trends of the materials in each category. For the sake of clarity, only
 142 results for a single incident heat flux, 50 kW m⁻², are presented at this stage. The influence of
 143 incident heat flux is presented in a later section. Two repetitions are presented for each material.

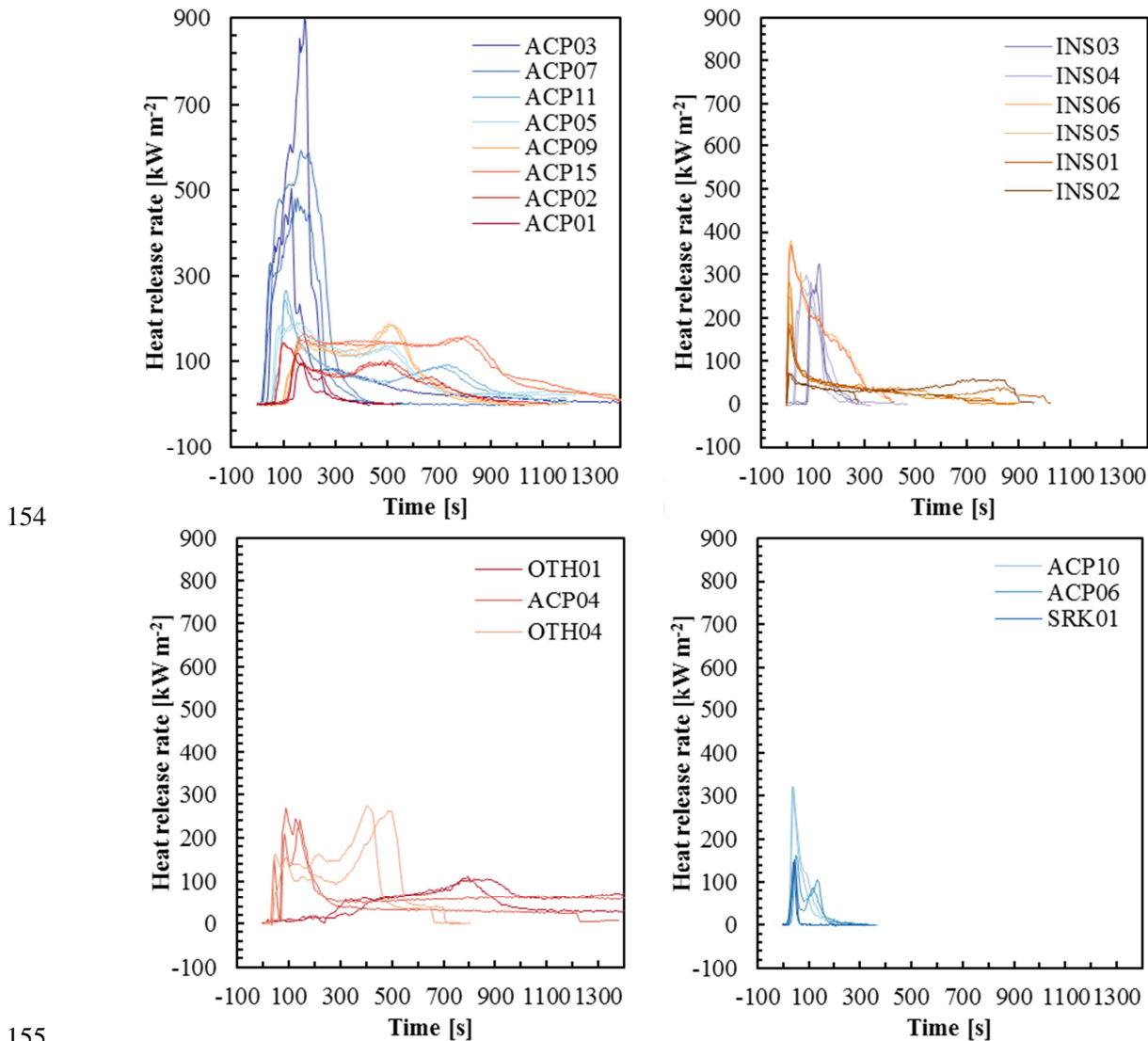
144 Table 1. Summary of basic categories of materials in the library.

Material description	Shorthand label	Sample IDs
ACPs with an organic core and may contain fire retardants, inorganic fillers, processing aides, etc.	ACPs	ACP01, ACP02, ACP03, ACP05, ACP07, ACP09, ACP11, ACP15
Materials with only a thin layer of organic material	Thin film	ACP06-S1, ACP10, SRK01-S1
Thermosets, generally aromatic compounds containing combinations of phenol resin and cellulose	Aromatics	ACP04, OTH01, OTH04
Insulation materials	Cellular	INS01, INS02, INS05, INS06
	Non-cellular	INS03, INS04

145

146 ACPs have a wide range of burning behaviour depending on their formulation (Fig. 1, top left).
 147 Materials which are pure or nearly pure thermoplastics behave as non-charring solids, such as
 148 ACP03 and ACP07. These are characterised by very rapid burning rates (500–900 kW m⁻² heat

149 release), no residue, and short burning durations (300 s). ACPs containing inorganic filler tend to
 150 have lower heat release ($100\text{--}275\text{ kW m}^{-2}$) and longer burning durations (400–1400 s). Some of
 151 the ACPs behave similar to charring materials as a char layer builds up and insulates the
 152 underlying virgin material from external heat. In general, the heat release for most of these
 153 materials is nonetheless relatively consistent throughout the duration of flaming.



156 Fig. 1. Heat release rate per unit area as a function of time for ACPs (top left), insulations (top
 157 right), aromatics (bottom left) and thin film samples (bottom right).

158 A number of key trends are noted for insulation materials (Fig. 1, top right). The total energy
 159 released by these materials, as indicated by the integral of the heat release, can be seen to be
 160 substantially lower than ACPs despite much larger sample thicknesses (up to 100 mm compared
 161 to up to 6 mm for ACPs). This is partially due to the low density of modern insulation materials
 162 leading to low sample mass. For cellular materials, the extremely low thermal inertia causes the
 163 surface to heat rapidly, and ignition is achieved quickly [11]. These have strong charring-like
 164 behaviour, with a relatively short peak heat release followed by significant decay for the

165 remaining duration of the test. Once a critical char depth is established, insufficient heat can
166 reach the virgin fuel to generate adequate pyrolysis gases, and hence extinguishment occurs.

167 The two non-cellular insulations – INS03 and INS04 – both undergo melting and otherwise
168 behave as non-charring thermoplastics. The melting leads to regression of the surface from the
169 heat flux of the cone, and thus a lower heat exposure is experienced. Consequently, the results
170 are somewhat lower than what may be deemed the true values. This behaviour, epitomised by
171 expanded polystyrene, is encountered regularly in the literature [12,13].

172 The aromatic formulations all generally behave as charring materials. The behaviour of the three
173 materials is still nonetheless quite varied, depending on the ratio of cellulose to phenol resin, and
174 on the binder used. The best performance is observed in OTH01 which is a high pressure
175 laminate with a phosphorous-based compound. Its higher density and inorganic additive lead to a
176 significantly longer ignition time, with a consistently low rate of heat release due to the lower
177 thermal inertia and increased charring respectively. The other two materials both ignite more
178 rapidly and have distinct peaks. From this, it is clear that the aromatic-based formulations have a
179 wide range of performance.

180 Finally, the heat release of materials where the organic component is only a thin layer are shown.
181 ACP06 and ACP10 consist of a thin layer of adhesive binding an aluminium profile to two
182 encapsulating aluminium sheets. The final material in this group, SRK01, is a weatherproof
183 membrane comprising a layer of polypropylene with an aluminium backing, and interwoven
184 glass fibre reinforcement. The materials in this category are somewhat distinct in their physical
185 geometries, but ultimately the thin layer of thermoplastic polymer leads to sharp heat release
186 before the material is quickly consumed and no fuel remains to sustain the flame.

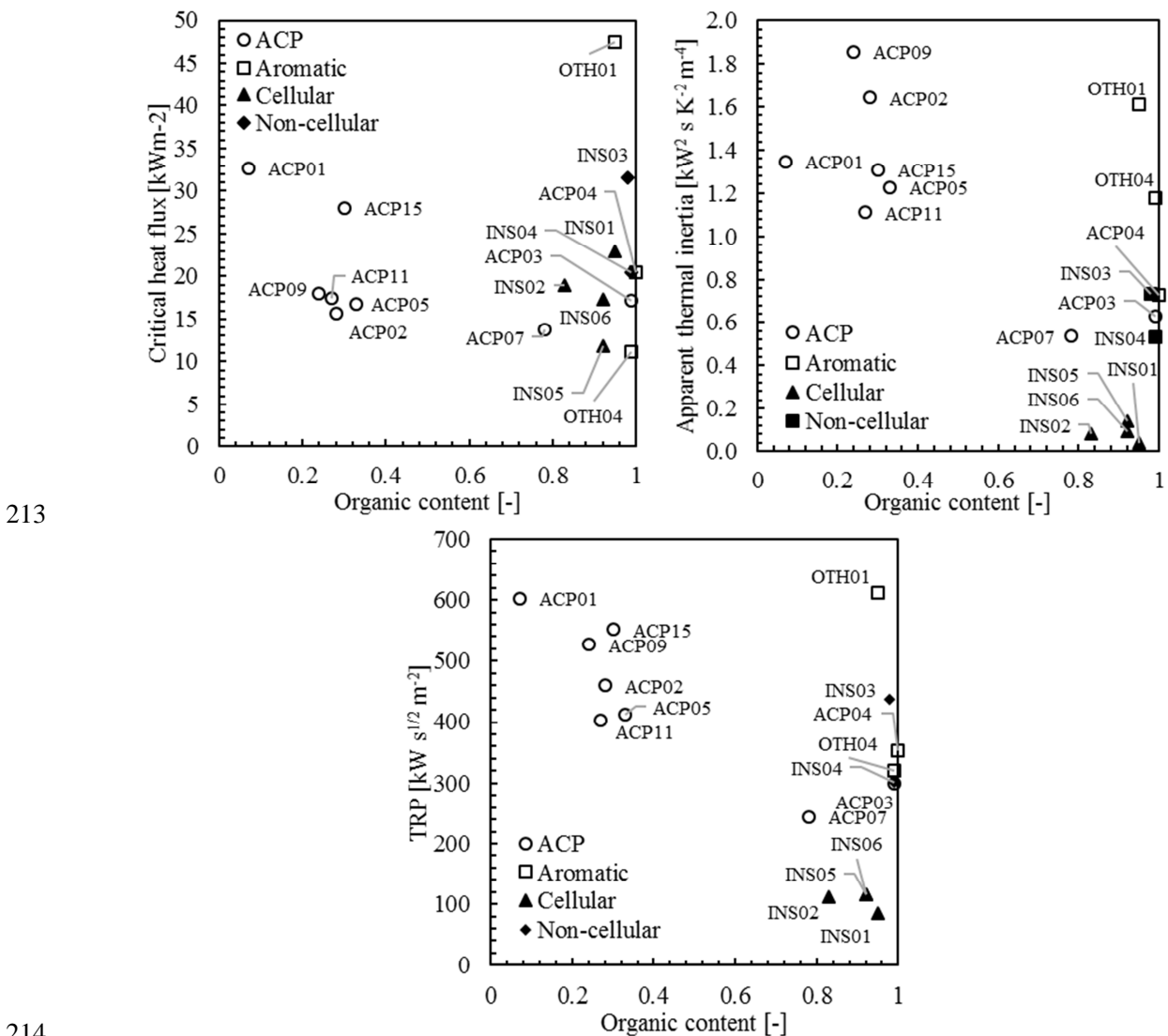
187

188 **5. Trends in ignition**

189 Ignition is a key risk for flame spread propagation. The two major parameters which affect the
190 time to ignition are the ignition temperature and the thermal inertia. The latter term is obtained
191 from cone calorimeter tests performed at different incident heat fluxes, and is a measure of how
192 rapidly a material heats up. For the case of materials with extremely low thermal inertias, such as
193 insulations, then the ignition is rapid regardless of the ignition temperature. The thermal response
194 parameter (TRP) is a function of these two terms (defined later in Eq. (5), and is thus used to
195 give a more generic assessment of the relative performance of different materials [14]. Each of
196 these parameters – critical heat flux, ignition temperature and apparent thermal inertia – is
197 plotted against organic content in Fig. 2 for all the materials.

198 The critical heat flux for the majority of ACPs lies within the range of 13–18 kW m⁻², and has
199 little to no dependence on the organic content expressed as a percentage (Fig. 2, top left).
200 Organic content here refers to carbon-based material, and includes materials such as polymers
201 and waxes. This is used in preference to terms such as “polyethylene content”, as this is only a
202 single example of a polymer, and there are likely to also be other minor components present such
203 as lubricants and processing aides which contribute to the equally flammability. One of the major
204 outliers for this trend is ACP15, which has predominantly Mg- and Ca-based inorganic
205 components, such as magnesium hydroxide, magnesium carbonate and calcium carbonate. X-ray
206 Diffraction (XRD) was used to characterise the inorganic composition of crystalline materials in

207 more detail [6]. This showed that compared to another similar ACP with Mg-based inorganics
 208 (ACP05), ACP15 had lower magnesium hydroxide, but higher calcium carbonate, and additional
 209 Si-based inorganics such as kaolinite, which is a form of clay, and quartz. These appear to
 210 significantly improve the critical heat flux (28.0 kW m^{-2} compared to 16.8 kW m^{-2} for ACP05)
 211 despite the similar total organic content, and this would be is beneficial to be able to resist
 212 ignition and flame spread.



214
 215 Fig. 2. Polymer content against critical heat flux for flaming ignition (top left), apparent thermal
 216 inertia (top right), and TRP (bottom).

217 There is a wide range of apparent thermal inertias, as calculated from the time to ignition across
 218 a range of heat fluxes in the cone (Fig. 2, top right). This is primarily due to the modern
 219 insulation materials, which have extremely low thermal inertia in order to reduce building energy
 220 usage and improve sustainability. Polyethylene normally has a reasonably low thermal inertia, as
 221 represented by ACP03 and ACP07, but this is substantially increased by the addition of
 222 inorganic components, as represented by the other materials in the ACP grouping. There is

223 however no linear trend, and all the materials with fire retardants or inorganic fillers have
224 increased thermal inertia. It is surprising that ACP01, which is predominantly inorganic and
225 would be expected to contain highly dense materials with high thermal conductivity, is within a
226 group with the other ACPs. The aromatics reasonably fall in the same ranges as ordinary ACPs.

227 TRP ultimately gives the best indication for resistance to ignition since it combines both
228 parameters (Fig. 2, bottom). Here, the ACPs at both the extreme ends of the range (ACP01 at
229 low polymer content and ACPs 03 and 07 at the high end) show distinct differences in
230 behaviour. The low thermal inertia of polyethylene results in a reduced TRP, where the critical
231 heat flux for the ACPs is similar. The extremely low thermal inertia of the foams is reflected in
232 the TRP, leading to very short ignition times, in the region of 1–10 s. The phenolic composite
233 OTH01 performs extremely well, with the highest TRP of all materials. This is despite the fact
234 that its composition is overwhelmingly organic. This highlights one of the issues in using the
235 organic content as a metric for assessing the viability of cladding materials. The reasonably high
236 TRP of INS03 is most likely an artefact of its melting behaviour, and in larger-scales it might not
237 be expected to perform so well. Overall, many of the ACPs have reasonable resistance to ignition
238 compared to other materials in the façade, such as insulation and various thermoplastics.

239

240 6. Trends in burning behaviour

241 The trends in the burning behaviour are shown for all the cladding materials. In each case, the
242 data is presented for an incident heat flux of 50 kW m^{-2} . The data was plotted for other heat
243 fluxes but there were no significant changes in the trends. Nonetheless, the effect of the incident
244 heat flux is covered in a subsequent section.

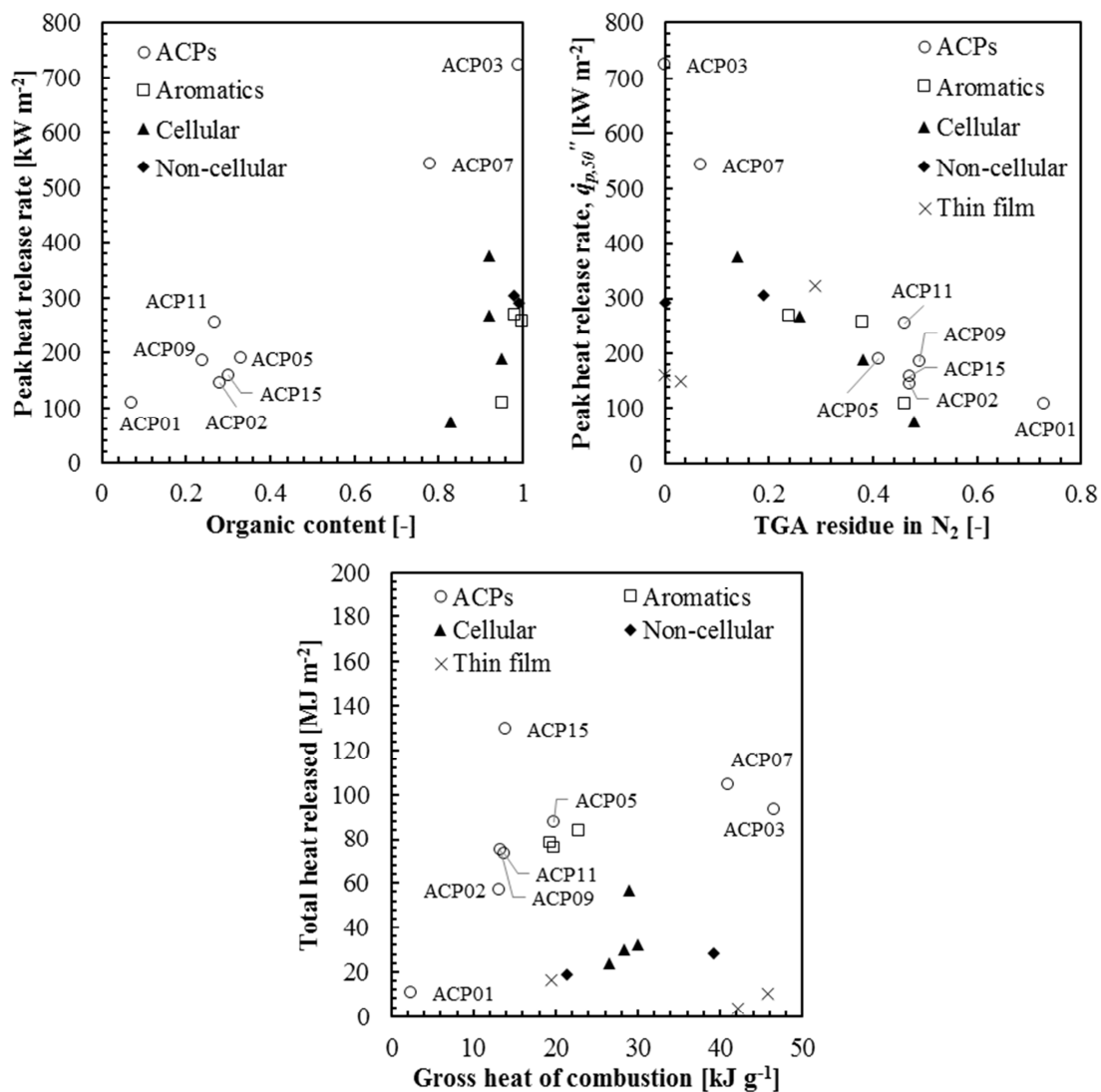
245 A reasonable trend in the organic content against peak heat release rate is observed for ACPs
246 (Fig. 3, top left). The organic content has been one of the common methods for rapid assessment
247 of cladding materials. There are still large disparities in the heat release for materials with
248 relatively low organic contents (mass fraction <0.35). The difference for ACP02 and ACP11 –
249 which both contain aluminium hydroxide in the same proportion but ACP02 includes vinyl
250 acetate in its polymer matrix – is in the order of a 100% increase for the peak heat release rate.
251 This 100% increase illustrates the problem with using organic content as an indicator of fire
252 performance.

253 Most notably however, the shortcoming of using the organic content can be seen for all the non-
254 ACP materials. The heat release rate for all these materials is significantly lower than the pure
255 thermoplastic ACPs (ACP03 and ACP07) despite the fact they have similar or the same organic
256 contents. The most extreme cases of this are for a phenolic composite (OTH01) and a foam
257 (INS02) where the heat release is only in the range of $75\text{--}125 \text{ kW m}^{-2}$. The coefficient of
258 determination, R^2 , of a linear trend for all materials is 0.168 but this is not plotted for clarity.

259 A more generalised correlation can be found by plotting the TGA residue at $800 \text{ }^\circ\text{C}$ in an inert
260 atmosphere against the peak heat release rate (Fig. 3, top right). The coefficient of determination
261 for the same materials as above (i.e. all except thin film) is improved to 0.653. This may be more
262 effective as a rapid screening evaluation, but still has limitations. This improvement in the
263 correlation is due to the fact that the propensity for char formation is taken into consideration.
264 This is beneficial for fire performance, as the heat feedback from the flame is inhibited by the

265 insulating properties of the char. This is mainly beneficial over long periods but is still well
 266 reflected in the peak heat release rate as the char layer is developing during the initial stages of
 267 heat exposure. The ACPs are shown to have the highest residues, most likely due to the very
 268 large quantities of fire retardant included to adequately improve their performance.

269 Overall, for a rapid initial assessment using low cost methods then it appears that the residue in
 270 an inert atmosphere is suitable for obtaining an estimation of the peak heat release of a material.
 271 TGA tests are more rapid than cone calorimeter tests, require less material from buildings, and
 272 require less sample preparation, thus requiring less labour. Nonetheless, the screening would be
 273 as a means to cross-reference to the fire performance and would not be adequate by itself.



274

275

276 Fig. 3. Flammability trends of organic content against peak HRR (top left), inert residue at 800
 277 °C against peak HRR (top right), and heat of combustion against the total heat released (bottom).

278 A breakdown in the trend can be seen for the thin film materials, and for one of the non-cellular
 279 insulations (Fig. 3, top right). For the thin films, the depth of the polymer layer is such that the

280 burning rate cannot reach a maximum steady rate of burning. The flame growth following
281 ignition is a feedback loop where heat from the flame increases the rate of volatiles generated,
282 which increases the flame heat flux. When the polymer layer is thin, then this process is not able
283 to reach a steady state and instead a decay phase initiates and extinction is reached rapidly.

284 Finally, the gross heat of combustion is shown against the total heat released (Fig. 3, bottom).
285 The gross heat of combustion is a parameter which gives the total possible energy that can be
286 released by a material under ideal conditions, and includes condensation of moisture vapour.
287 While the net heat of combustion may be more pertinent, the gross heat of combustion is used in
288 this comparison for simplicity. There appears to nonetheless be little correlation between the
289 gross heat of combustion and the total heat released, suggesting that there are a large number of
290 other phenomena which are complicating the analysis. One of these is likely to be combustion
291 inefficiency, where insufficient oxidiser can reach the fuel surface in the cone calorimeter test.
292 The rate of energy released is then reduced compared to the pure oxygen conditions in the bomb
293 calorimeter. Another complication is the thickness of different materials in the cone testing,
294 leading to higher total heat released when assessed per unit area. The heat transfer conditions
295 will change depending on the thickness of the material e.g. whether or not the material is
296 thermally thin or not, and thus more heavily influenced by boundary conditions.

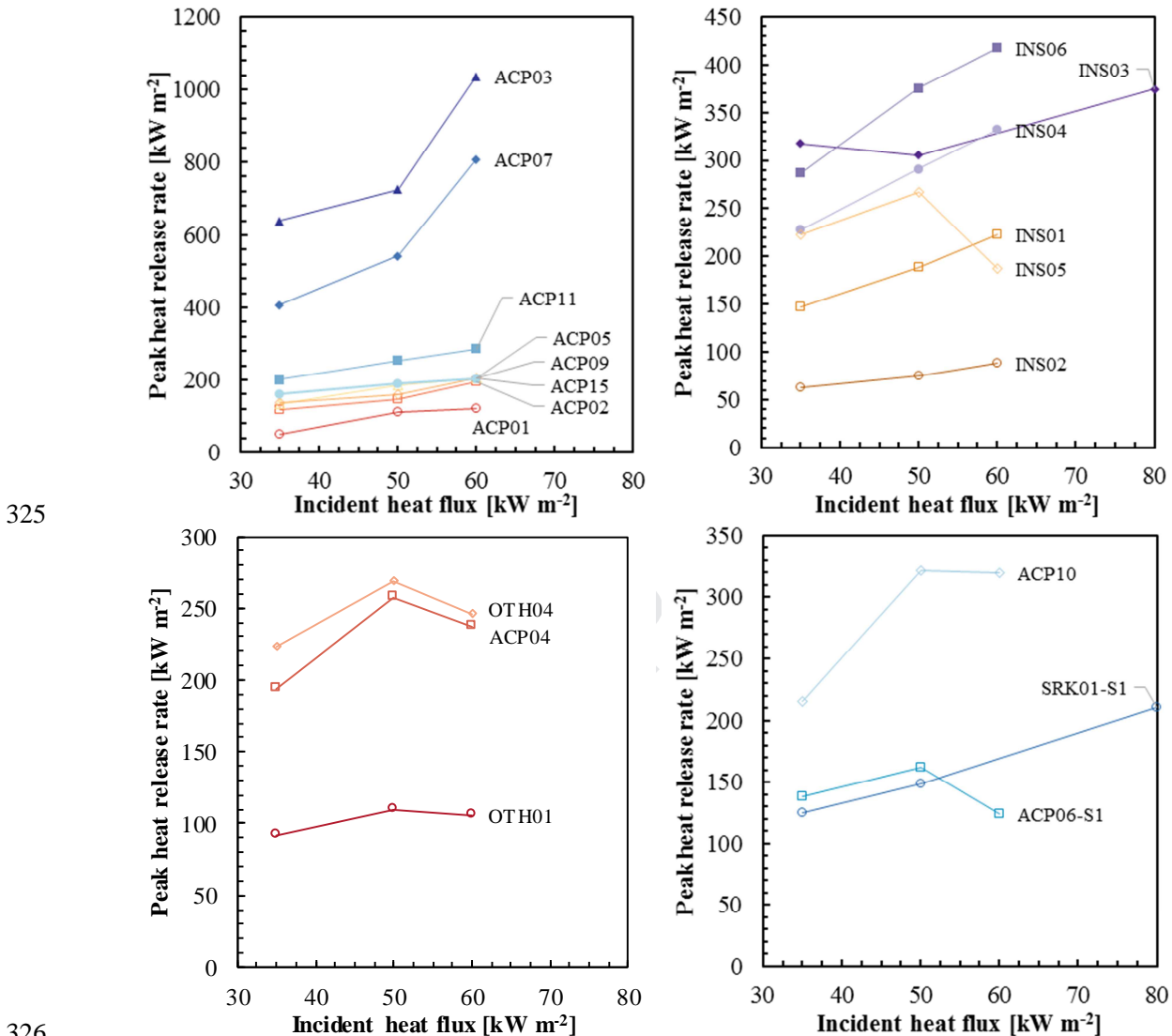
297 The change in incident heat flux can lead to differences in the burning behaviour depending on
298 the type of material. This behaviour is well known for many polymer-based materials [15]. The
299 results of peak heat release rate for all materials at three incident heat fluxes – 35, 50 and 60, or
300 80 kW m⁻² in some rare cases – are shown in Fig. 4.

301 The change in peak heat release rate (PHRR) is relatively constant for the ACPs with inorganic
302 components. Many of the materials converge on the same PHRR (~200 kW m⁻²) when exposed
303 to 60 kW m⁻². The significantly higher PHRR for thermoplastics at 60 kW m⁻² exposure is
304 concerning since it suggests higher burning rates in the worst cast scenarios. Higher heat release
305 rates increase the flame length which in turn increases the length preheated by the flame. This
306 may lead to accelerating flame spread velocities which can propagate rapidly up a building.

307 The results of the polyester wool insulation, INS03, are poor due to the melting and regression of
308 the surface. Furthermore, the material is highly heterogeneous leading to inconsistencies in the
309 peak heat release rate. In contrast, the superior manufacturing quality and uniformity of
310 expanded polystyrene, INS04, shows a more consistent trend despite the same difficulties with
311 melting and surface regression. One of the polyurethane-based polyisocyanurate (PIR) cellular
312 foams, INS05, shows a substantial decrease when exposed to higher heat fluxes. The formulation
313 of PIR foams is highly complicated, and often a plateau in the behaviour is noted at moderate to
314 high heat fluxes [16]. This is due to the extremely low thermal inertias, the chemical
315 composition, and the formation of a protective char. At higher heat fluxes, the char is generated
316 more rapidly and less pyrolysis gases released in the initial stages of exposure. The other
317 insulation materials otherwise show a typical relationship with the incident heat fluxes. The
318 highest heat release rates are consistently recorded for INS06, which is a rigid polyurethane
319 foam.

320 The PHRR of aromatics shows little dependence in the incident heat flux. The charring nature of
321 the materials means that the char layer which forms will begin to reduce the heat flux from the
322 cone heater received by the material in-depth. As a result, heat is propagated by the hot char at a

323 rate which is not heavily influenced by the external heat. A change in oxygen conditions, i.e.
 324 increased oxidiser flow, may however rapidly increase the burning rate.



327 Fig. 4. Peak heat release rate for various incident heat fluxes in the cone calorimeter – ACPs (top
 328 left), insulations (top right), aromatics (bottom left), and thin films (bottom right). It should be
 329 noted that the magnitude of the ordinate axis is not consistent between plots.

330 Consistent behaviour is recorded in the PHRR of the weatherproofing membrane, SRK01 due to
 331 the high quality manufacturing process which guarantees a constant thickness of material. For
 332 the other two thin films, ACP06 and ACP10, then the adhesive is not constant along the
 333 thickness of a panel, or between panels. The change in resin quantity is known from the mass
 334 measurements of each sample, and thickness measurements using digital callipers. The net result
 335 of this is that there is large variability in the results. This means that the performance is dictated
 336 more by the quantity of resin on a specific panel than by the incident heat flux. For ACP10, the
 337 estimated resin mass was approximately in the order of 1–2 times higher in 50 kW m⁻² than the
 338 other heat fluxes. Only a limited number of samples were received, and thus this error could not

339 easily be eliminated. These materials are a clear example where additional work is required but
 340 which are completely neglected. They are common replacement materials for “flammable”
 341 cladding (e.g. ACP03) but there is no assessment of their fire performance. These materials can
 342 only be used if their performance is known and quantified, so the fire risk can be assessed.

343 7. Model application

344 Ignition and heat release are two fundamental aspects in flame spread, particularly for vertical or
 345 wind-aided flame spread. The application of a simple model allows insight into the potential
 346 flame spread performance of cladding materials by accounting for the combined contributions of
 347 the ignitability and the burning behaviour.

348 The steady flame spread velocity for concurrent flow conditions for a thermally thick solid is
 349 given by Eq. (1) [17]:

$$v_p = \frac{4 \dot{q}_f''^2 (x_f - x_p)}{\pi k\rho c (T_{ig} - T_\infty)^2} \quad (1)$$

350 Where v_p is the spread velocity, \dot{q}_f'' is the heat from the flame over the pyrolysis length $x_f - x_p$,
 351 the term x_f is the flame height, x_p is the pyrolysis height, the lumped parameter $k\rho c$ is the
 352 thermal inertia, T_{ig} is the ignition temperature and T_∞ is the ambient temperature. Quintiere *et al.*
 353 [18] found the following relationship for the flame height against a vertical wall (Eq. (2):

$$x_f = k_f \dot{Q}_f'' x_p \quad (2)$$

354 Where k_f is an empirical constant and \dot{Q}_f'' is the heat release per unit area of the material. The
 355 latter is obtained from the cone calorimeter testing, using the peak heat release rate for tests with
 356 an incident heat flux of 50 kW m⁻². Substituting Eq. (2) into Eq. (1) yields Eq. (3):

$$\frac{v_p}{x_p} = \frac{4 \dot{q}_f''^2 (k_f \dot{Q}_f'' - 1)}{\pi (T_{ig} - T_\infty)^2 k\rho c} \quad (3)$$

357 Which gives an estimate of the flame spread velocity for a given pyrolysis height. This contains
 358 two parameters, among others, which have been drawn out by other authors. Quintiere *et al.* [18]
 359 highlighted the significance of the a parameter in describing the contribution of the heat release
 360 of material to the upward flame spread propagation, Eq. (4):

$$a = k_f \dot{Q}_f'' - 1 \quad (4)$$

361 While Tewarson [14] has given the thermal response parameter (*TRP*) as a fundamental material
 362 property to describe ignitability and hence subsequent flame propagation, Eq. (5):

$$TRP = (T_{ig} - T_\infty) \sqrt{k\rho c} \quad (5)$$

363 A critical condition can further be highlighted as the ‘ a ’ parameter tends to zero. Rearranging
 364 Eq. (4) for $a = 0$ then gives an estimation of the critical heat release rate required to maintain
 365 positive, i.e. accelerating, flame spread. This is found to be $\dot{Q}_f'' = 100\text{--}160$ kW m⁻² for values of
 366 $k_f = 0.006\text{--}0.01$ kW⁻¹ m². Thus, in cases where the heat release is sufficiently small then
 367 upward flame spread is predicted to not occur.

368

369

Table 2. Summary of computed properties relevant to vertical flame spread.

ID	Category	a - Eq. (4)	TRP $\text{kW s}^{1/2} \text{m}^{-2}$ Eq. (5)	v_p/x_p s^{-1} Eq. (3)
INS02	Cellular	-0.25	112	-15.58
OTH01	Aromatic	0.09	613	0.19
ACP01	ACP	0.10	602	0.21
ACP15	ACP	0.59	552	1.55
ACP02	ACP	0.45	459	1.71
ACP09	ACP	0.85	527	2.44
ACP05	ACP	0.90	413	4.18
ACP06-S1	Thin film	0.62	307	5.20
ACP11	ACP	1.55	402	7.62
INS03	Non-cellular	2.05	437	8.56
ACP04	Aromatic	1.57	352	10.05
OTH04	Aromatic	1.68	320	13.04
INS04	Non-cellular	1.91	301	16.71
SRK01-S1	Thin film	0.49	115	29.21
ACP10	Thin film	2.22	243	29.88
ACP03	ACP	6.25	300	55.05
ACP07	ACP	4.43	244	59.43
INS01	Cellular	0.88	85	98.51
INS05	Cellular	1.67	116	98.94
INS06	Cellular	2.76	115	165.94

370

371 These properties have ultimately been calculated and summarised for all the cladding materials
372 studied (Table 2). For this, k_f has been taken to be $0.01 \text{ kW}^{-1} \text{ m}^2$, and \dot{q}_f'' has been assumed a
373 constant value of 25 kW m^{-2} [17,18], although values in the literature vary from $20\text{--}60 \text{ kW m}^{-2}$
374 [19,20]. The aim of this is a comparison, and thus the selection of these constants is arbitrary.
375 The heat release of the material, \dot{Q}_f'' , was taken from the peak heat release rate in the cone, as
376 described earlier. This will have a tendency to overestimate the flame spread in all cases, but will
377 be particularly conservative for the case of charring materials with high thermal inertia, namely
378 the cellular insulation foams. This is due to the fact that their peak heat release is short, while the
379 burning time is much longer as the char layer builds up. Later literature [12,21] has suggested
380 that an average peak should be taken but there is not a clear definition of what form this should
381 take. This would thus require assumptions about which timescale should be selected, and also
382 how to best eliminate the effects of smouldering which will contribute heat release in the later
383 stages for some materials. For this reason, the conservative use of peak heat release rate for \dot{Q}_f'' is
384 maintained.

385 A negative parameter is predicted for INS02, a phenolic foam insulation. Other ACPs and
386 OTH01 also have values close to zero (i.e. <1.0), suggesting little potential for flame spread. On
387 the other end of the scale, the thermoplastics and the insulation foams with very low thermal
388 inertias are expected to have rapid rates of spread. This is based on the very high heat release
389 rates leading to large flame heights for the thermoplastics, and rapid surface heating for the
390 insulations due to their thermal inertia. These predictions have an apparent good qualitative
391 agreement with results obtained in vertical Lateral Ignition and Flame spread Tests (LIFT).
392 However, until detailed analysis is performed on the vertical flame spread then a full comparison
393 and assessment cannot be made.

394 The two ACPs with only aluminium hydroxide as the filler, ACP02 and ACP11, show vastly
395 different behaviour. This suggests that the addition of vinyl acetate in ACP02 would significantly
396 inhibit the propensity for flame spread. The magnesium hydroxide-based ACPs – ACP15,
397 ACP09 and ACP05 – all suggest little spread potential but there is a wide range for these
398 materials. The aromatics – ACP04, OTH01 and OTH04 – show a wide range of potential flame
399 spread. The phosphorous-based flame retardant, high density and high phenol resin content of
400 OTH01 contribute to very low heat release and a high charring yield. Conversely, both ACP04
401 and OTH04 contain no retardants and may be expected to spread flame in the event of a fire.

402 Ultimately, these materials are included as part of a system and there will be complicated
403 interactions between the different elements. Some work has been performed [22] to investigate
404 the interactions between different ACPs and insulations separated by a cavity, which has also
405 been represented by the seven full-scale tests commissioned by the Government in the UK [23].
406 The intermediate-scale tests gave the opportunity to identify the contribution of individual
407 components, which can be tied into the Cladding Materials Library data and the analysis here.
408 Qualitatively, the results presented in the simple model here are in agreement with the behaviour
409 in the intermediate-scale tests, the large-scale tests commissioned in the UK, and large-scale tests
410 performed as part of this project. More thorough analysis is required and significantly more
411 testing is required to robustly connect material-level and full-scale façade fire behaviour.

412 Through methodical and thorough understanding of the materials and systems used, the eventual
413 development of simple models may be possible. This allows engineers to adequately design
414 façade systems, instead of using arbitrary metrics such as polyethylene content which provide no
415 evidence of performance. Optimisation of buildings by balancing a flammability index with
416 other design objectives [24] will enable more efficient and effective buildings in the future.

417

418 8. Conclusions

- 419 • The cladding crisis encompasses all materials in the system, and is not limited to ACPs.
420 The organic content is shown to be a poor indicator of fire performance and ignitability
421 for ACPs, and has no application to non-ACPs, which account for around half of the
422 materials. For rapid initial screening, the TGA residue in nitrogen appears to offer some
423 improvement in the correlation with peak heat release rate. TGA tests are rapid, require
424 little material from buildings, and require relatively little sample preparation.
- 425 • A systematic assessment of the fire behaviour of different components in a cladding
426 system have been shown and described. The different components are shown to have

- 427 very different performances, and even within a single category there is a wide range of
 428 performance.
- 429 • A simple model to inform on potential vertical flame spread identified materials likely to
 430 undergo rapid spread rates and those where spread is less likely to occur. Some phenolic-
 431 based materials and heavily retarded charring-forming materials were indicated to not
 432 promote spread, while pure thermoplastics, cellular insulation and some thin film
 433 organics were expected to perform poorly. There is apparent good agreement with the
 434 vertical flame spread LIFT results but this requires in-depth analysis of the flame spread
 435 results to be certain.
 - 436 • Further work is required to understand the interaction between materials and effects
 437 observed at larger-scales. Material-level analysis acts as an initial step in building
 438 knowledge, and ensuring maximum value can be gained from expensive full-scale tests.
 439

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446

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516 **Figure captions**

517

518 Fig. 1. Heat release rate per unit area as a function of time for ACPs (top left), insulations (top
519 right), aromatics (bottom left) and thin film samples (bottom right).

520 Fig. 2. Polymer content against critical heat flux for flaming ignition (top left), apparent thermal
521 inertia (top right), and TRP (bottom).

522 Fig. 3. Flammability trends of organic content against peak HRR (top left), inert residue at 800
523 °C against peak HRR (top right), and heat of combustion against the total heat released (bottom).

524 Fig. 4. Peak heat release rate for various incident heat fluxes in the cone calorimeter – ACPs (top
525 left), insulations (top right), aromatics (bottom left), and thin films (bottom right). It should be
526 noted that the magnitude of the ordinate axis is not consistent between plots.

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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